



# Third-order optical nonlinearities and high-order harmonics generation in Ni-doped CsPbBr<sub>3</sub> nanocrystals using single- and two-color chirped pulses

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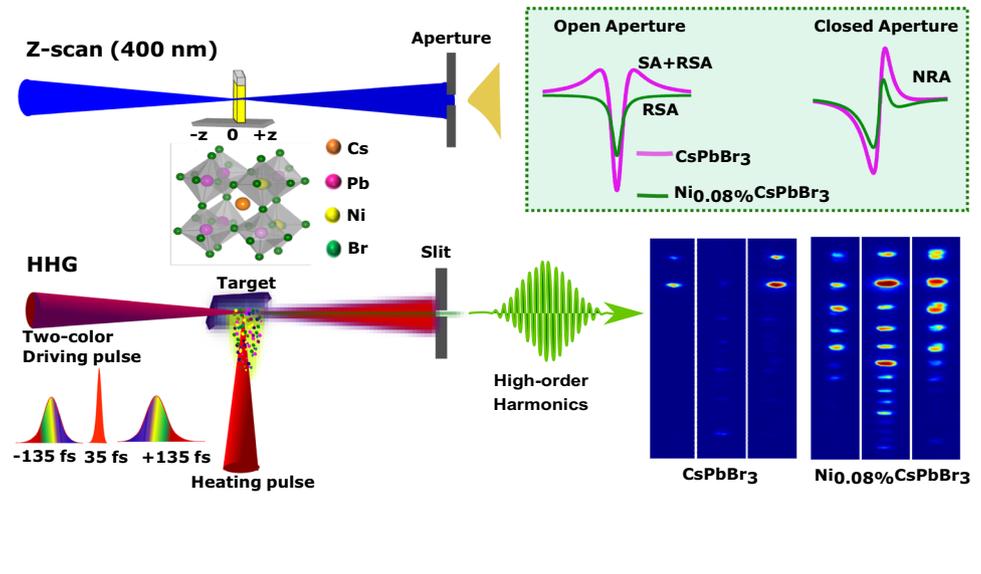
## ABSTRACT

Organic–inorganic lead halide perovskite nanocrystals (NCs) have received the phenomenal attention of researchers in the last few years due to their advanced optical properties and developments in simple synthetic procedures. In this work, we demonstrate the effect of Ni-dopant on the ultrafast third-order nonlinear optical (NLO) properties of two-dimensional (2D) CsPbBr<sub>3</sub> NCs using 60 fs laser pulses at a wavelength of 400 nm. Based on the probe pulse intensity, pure CsPbBr<sub>3</sub> NCs exhibited saturable absorption and reverse saturable absorption (RSA). The doping of NCs by Ni at different concentrations significantly modified the nonlinear absorption mechanism where the RSA was followed by the SA with growing input laser intensity. The nonlinear refractive index of Ni-doped CsPbBr<sub>3</sub> NCs decreased when compared with the pure CsPbBr<sub>3</sub>. In addition, the NLO properties of these 2D NCs are correlated with the efficiency of the high-order harmonics generated during propagation of the two-color (800 + 400 nm) and single-color (800 nm) chirp-free (35 fs) and negatively/positively chirped pulses through the laser-induced plasmas produced on the surface of undoped and Ni-doped perovskite NCs. The ablation of Ni-doped NCs allowed increasing the harmonic yield compared to undoped CsPbBr<sub>3</sub> NCs. Our studies undoubtedly demonstrate the advanced lower and higher order NLO properties of Ni-doped 2D CsPbBr<sub>3</sub> NCs.

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## GRAPHICAL ABSTRACT



## Introduction

Lead halide-based perovskite nanocrystals (NCs) have unique and venerable photophysical properties, which permitted their efficient application in various optoelectronic devices such as photodetectors, light-emitting diodes, solar cells, and lasers. Their optical, optoelectronic, photonic, and third-order nonlinear optical (NLO) properties were frequently reported and reviewed [1–16]. The excited-state dynamics (pumped with 400 nm), NLO, and photoluminescent properties of the undoped and Ni-doped CsPbBr<sub>3</sub> NCs were explored in our recent work [17]. It was demonstrated that the photoluminescence (PL) yields from Ni-doped CsPbBr<sub>3</sub> NCs were higher than from pure CsPbBr<sub>3</sub> NCs. Further, the central wavelength was observed to vary in the PL spectra depending on the weight percentage of the dopant. In addition, a few research groups have explored the outstanding NLO properties of different perovskite nanomaterials such as (i) saturable absorption (SA) (ii) reverse saturable absorption (RSA) (iii) two-photon absorption (2PA) (iv) multiphoton absorption and (v) nonlinear refraction using different excitation wavelengths and input pulse durations [18–33]. However, there are very few reports of the NLO properties of these exotic molecules in the UV spectral region. Additionally,

the high-order harmonics generation (HHG) in solids, gases, and laser-induced plasmas (LIPs) attracted great attention due to the necessary information of the efficient sources of coherent extreme ultraviolet (XUV) radiation [34–57]. Various capable solid materials and gas jets with control of driving pulse duration to obtain better conversion efficiencies and cut-off of harmonics were analyzed [40, 58–64]. Recently, Hussain et al. [65] have demonstrated HHG with spectral shifts in the obtained harmonics in silicon (Si) and zinc oxide (ZnO) achieved by scanning the focal position of the 85 fs driving pulses at a wavelength of 2.123  $\mu\text{m}$ . The variations of the spectral shape and harmonic distribution from silver plasma using chirped pulses from the 793 nm, 44 fs laser pulses were also reported [66]. Those studies underlined the role of chirped pulses in enhancing the harmonic's emission. Surprisingly, there are hardly any reports on the understanding of the correlation between different NLO properties of these NCs, for instance, the HHG yield and the real and imaginary parts of nonlinear susceptibilities.

In this study, we report the NLO properties of undoped and Ni-doped CsPbBr<sub>3</sub> NCs using femtosecond pulses at the wavelength ( $\lambda$ ) of 400 nm. We also analyze the high-order harmonics spectra using the positively and negatively chirped femtosecond laser pulses propagating through the plasmas

generated on the NCs surfaces. A correlation between the low-order NLO properties and the HHG in these NCs is discussed. The examination of innovative and effective materials (ablated Ni-doped NCs in the current case) in this way will allow achieving XUV radiation for a diversity of applications in attosecond physics and nonlinear spectroscopy [67–70].

## Experimental details

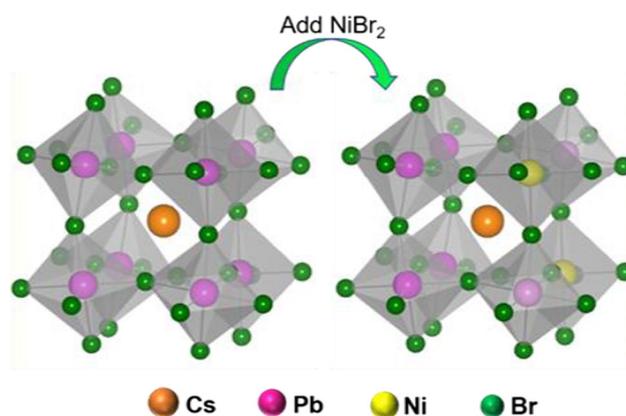
### Synthesis and characterization of nanocrystals

1 mM of  $\text{PbBr}_2$  and 0.08 mM  $\text{NiBr}_2$  were loaded into a three-neck round bottom flask containing 5 ml of 1-octadecene with 0.5 ml of dried oleic acid and dried oleylamine (here, oleic acid and oleylamine act as capping ligands). This was degassed for about one hour under vacuum ( $10^{-2}$  mbar) at 120 °C. The flask was subsequently flushed with nitrogen and heated to 150 °C. Later the temperature was raised to 210 °C and was kept for 5–10 minutes to reach the temperature. Subsequently, the cesium oleate solution was heated to the chosen reaction temperature (typically 150 °C) before 5 ml of 1 mM of cesium oleate (1 mM of  $\text{CsCO}_3$ , 20 ml of 1-octadecene, and 1.25 ml of oleic acid) solution was swiftly injected into the prepared precursor solution. After five seconds, the reaction mixer was cooled under an ice bath. Subsequently, we obtained the precipitated NCs by adding them with 5 ml of acetone.

The synthesis of Ni-doped  $\text{CsPbBr}_3$  was confirmed by observing the change in NCs formation as both the Ni:Pb ratio vary during the synthesis [71]. Post-synthesis cation exchange procedures can incorporate divalent guest cations into  $\text{CsPbBr}_3$  NCs, according to our findings. The aliovalent exchange of  $\text{Cs}^+$  by smaller  $\text{Ni}^{+2}$  cations can substantially destabilize the perovskite structure significantly, necessitate charge compensation. As a result, we assumed that the solely isovalent exchange of  $\text{Pb}^{+2}$  by  $\text{Ni}^{+2}$  occurred. Chemical mapping was used to validate the incorporation and distribution of the guest cation using energy-dispersive X-ray spectroscopy (EDS) data [72]. Schematic illustration of the structural properties of  $\text{CsPbBr}_3$  NCs before and after substitution of  $\text{Pb}^{+2}$  with 0.08 mM of  $\text{Ni}^{+2}$  ion is shown in Fig. 1.

Our previous works [17, 73] discussed the UV–Visible absorption spectra, PL characteristics, and

surface morphologies of pristine ( $\text{CsPbBr}_3$ ) NCs, 0.03 percent Ni, 0.05 percent Ni, 0.08 percent Ni, 0.1 percent Ni-doped  $\text{CsPbBr}_3$  NCs using transmission electron microscopy (TEM) pictures. In a nutshell, the absorption spectra of  $\text{CsPbBr}_3$  and  $\text{NiCsPbBr}_3$  were measured in the 200–800 nm region using a UV–Visible–NIR spectrophotometer (Model UV 3600, M/s Shimadzu). The samples were organized by diluting 20  $\mu\text{L}$  of the crude NCs solution in  $\sim 1$  mL of toluene and placed in quartz cuvettes (1 cm).  $\text{CsPbBr}_3$  and  $\text{NiCsPbBr}_3$  had absorption peaks near 482 and 508 nm. The emission spectra were recorded using a spectrofluorometer (from Horiba Jobin Yvon) and the corresponding PL emission peaks were observed at 530 and 566 nm. Ni-doping did not introduce a new absorption band in  $\text{CsPbBr}_3$  and the resultant excitation absorption of the  $\text{NiCsPbBr}_3$  NCs shifts to the red spectral region. The lifetime of photogenerated charge carriers was determined by time-resolved photoluminescence.  $\text{NiCsPbBr}_3$  had a longer lifetime (13.41 ns) compared to its pristine counterpart (11.38 ns), which can be considered as an advantage for application in LED devices [17]. The synthesized NCs had a square-like shape with an average size of  $\sim 20$  nm, according to the TEM studies. These NCs possessed a well-defined crystalline structure; it was observed in the orthorhombic phase. The size and uniformity of attained NCs were influenced by the doping of Ni. The measured lattice fringe spacing for pure and 0.08 percent Ni was 0.394 nm and 0.390 nm, respectively, indicating that Ni-doped  $\text{CsPbBr}_3$  NCs had a restricted size distribution.



**Figure 1** Schematic illustration of the structural properties of  $\text{CsPbBr}_3$  NCs before and after substitution of  $\text{Pb}^{+2}$  with 0.08 mM of  $\text{Ni}^{+2}$  ion.

### Z-scan measurements

The Ti: sapphire (Spectra-Physics, Model Spitfire Ace) laser system used in the present studies delivered linearly polarized 800 nm, 35 fs, 1 kHz pulses. The laser pulses were broadened to 60 fs before the focusing lens (200 mm focal length, beam waist radius  $w_o = 16 \mu\text{m}$ ) along the Z-scan path. The second harmonic (400 nm) of 800 nm radiation was generated using a 0.2 mm thick barium borate (BBO, type I) crystal and the 800 nm pulses were filtered using the color filter [Fig. 2(a)]. The pulse energies of 0.1  $\mu\text{J}$  (peak intensity  $(I_o) = 2.6 \times 10^{11} \text{ W/cm}^2$ ) and 0.15  $\mu\text{J}$  ( $I_o = 3.9 \times 10^{11} \text{ W/cm}^2$ ) were used to analyze the third-order NLO properties of the studied samples. The 1-mm thick fused silica cell filled with NCs suspension possessing a concentration of typically  $4 \times 10^{-4}$  mM was moved along the Z-axis of the focused laser beam as shown in Fig. 2a. For open-aperture (OA) and closed-aperture (CA) Z-scan measurements, after propagation of the sample, the two beams were collected by photodiodes 1 and 2, respectively. In the case of CA, the aperture was adjusted to allow the propagation of  $\sim 15\%$  of the input beam. At each point, the photodiode signal versus position of the sample was monitored by a personal computer using a data acquisition program developed using LabVIEW.

### HHG experimental layout

Figure 2b illustrates the schematic of the experimental setup used for HHG measurements. We used a Ti: Sapphire laser (800 nm, 35 fs) operated at 100 Hz pulse repetition rate. The 200 ps, 800 nm heating pulses (HP) from the same laser ablated the 2D NCs targets to create the plasma plumes. The harmonics spectra were obtained using the chirp-free single-color pump (SCP, 800 nm wavelength) and two-color pump (TCP, 800 nm + 400 nm wavelengths) of the LIPs produced on the surfaces of the studied samples. We also used the negatively and positively chirped 135 fs radiation from this laser as the driving pulses (DP) and compared the harmonic spectra generated using the above pulses with those produced by the chirp-free 35 fs pulses. The intensities of the chirp-free and chirped pulses in the focal plane were  $4 \times 10^{14}$  and  $0.9 \times 10^{14} \text{ W/cm}^2$ , respectively. The chirp of the DPs was varied by changing the separation of the gratings in the pulse compressor of Ti:

sapphire laser [66]. The chirp-free, negatively, and positively chirped pulses contain similar spectral components while showing a difference in their temporal characteristics and distribution of the spectral components along the laser pulse (see also Sect. 3.2). The details and description of the 2D NCs samples preparation for HHG measurements are presented in our earlier work [73]. Briefly, the NCs were dispersed in toluene solvents and then precisely drop-casted on the glass substrate until drying. The approximated length  $\times$  width  $\times$  thickness of NCs target was  $\sim 8 \text{ mm} \times 6 \text{ mm} \times 1 \text{ mm}$ . The substrates with NCs were placed in the target chamber (Fig. 2b) and ablated by HPs. The DPs propagated through the plasma plumes approximately 0.2 mm above the target surface. The emitted harmonics from LIPs were directed to the XUV chamber, which consists of a gold-coated cylindrical mirror (CM), flat field grating (FFG), and microchannel plate (MCP). Finally, the harmonic spectra were recorded using a charge-coupled device camera (CCD).

## Results and discussion

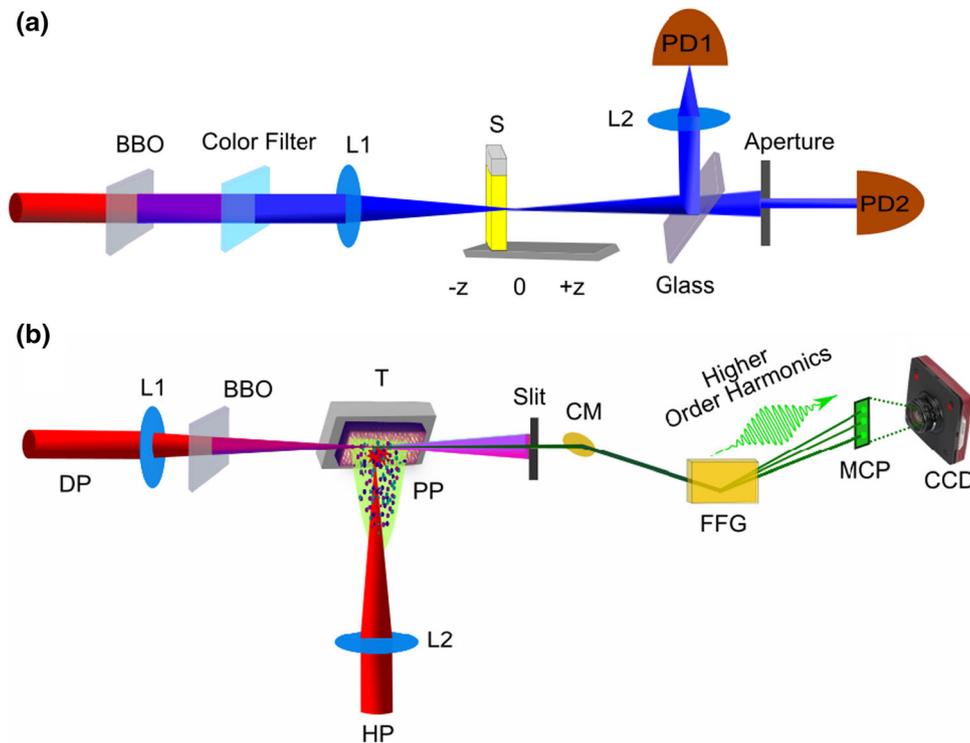
### Third-order NLO properties of nanocrystals

The third-order NLO properties of five (CsPbBr<sub>3</sub> and Ni (0.03, 0.05, 0.08, and 0.1 %)-doped CsPbBr<sub>3</sub> NCs) samples are measured using the standard Z-scan technique at wavelength ( $\lambda$ ) of 400 nm. From the OA and CA Z-scan measurements, one can obtain the nonlinear absorption coefficient ( $\beta$ ), saturation intensity ( $I_{\text{sat}}$ ), and nonlinear refractive index ( $\gamma$ ) of these species. The normalized transmittance of studied samples in the case of RSA and combination of SA and RSA can be fitted by the following equations [74]

$$T_{\text{RSA}}(z) \approx 1 - \frac{q}{2\sqrt{2}} \tag{1}$$

$$T_{\text{SA+RSA}}(z) \approx \left(1 - \frac{q}{2\sqrt{2}}\right) \times \frac{I_o}{I_{\text{sat}}(x^2 + 1)} \tag{2}$$

Here  $q = \beta I_o L_{\text{eff}} / (1 + z^2 / z_o^2)$ ,  $x = z / z_o$ ,  $z_o = k (w_o)^2 / 2$  is the Rayleigh length.  $w_o$  is the beam waist radius,  $k = 2\pi / \lambda$  is the wavenumber,  $I_o$  is the intensity of the laser beam in the focal plane.  $L_{\text{eff}} = [1 - \exp(-\alpha_o L)] / \alpha_o$  is defined as the effective length of the samples,  $\alpha_o$  is the linear absorption coefficient of sample, and  $L$  is the thickness of the sample.



**Figure 2** Experimental layouts for Z-scan and HHG measurements. **a** Z-scan scheme. BBO: barium borate crystal; L1: focusing lens ( $f=200$  mm), S: sample (1-mm thick cell filled with NCs suspension); L2: focusing lens for the collection of the propagated beam ( $f = 200$  mm); PD1 and PD2: photodiodes for the measurements of propagated radiation in the case of open-aperture and closed-aperture schemes, respectively. **b** HHG

scheme. DP: 800 nm driving pulses; L1: focusing lens ( $f = 500$  mm) for the driving pulses; BBO: barium borate crystal; T: target; PP: plasma plume; L2: focusing lens ( $f = 200$  mm for the heating pulses; HP: 200 ps, 800 nm heating pulses; CM: gold-coated cylindrical mirror; FFG: flat field grating; MCP: microchannel plate; CCD: charge-coupled device camera.

The normalized transmittance in the presence of nonlinear refraction and absorption (NRA) can be written as [75]

$$T_{\text{NRA}}(z) = 1 + \frac{2(-\rho x^2 + 2x - 3\rho)}{(x^2 + 9)(x^2 + 1)} \Delta\Phi_0 \quad (3)$$

Here  $\Delta\Phi_0 = k\gamma I_0 L_{\text{eff}}$  and  $\rho = \beta/2k\gamma$ . Correspondingly, the nonlinear refraction index can be determined as

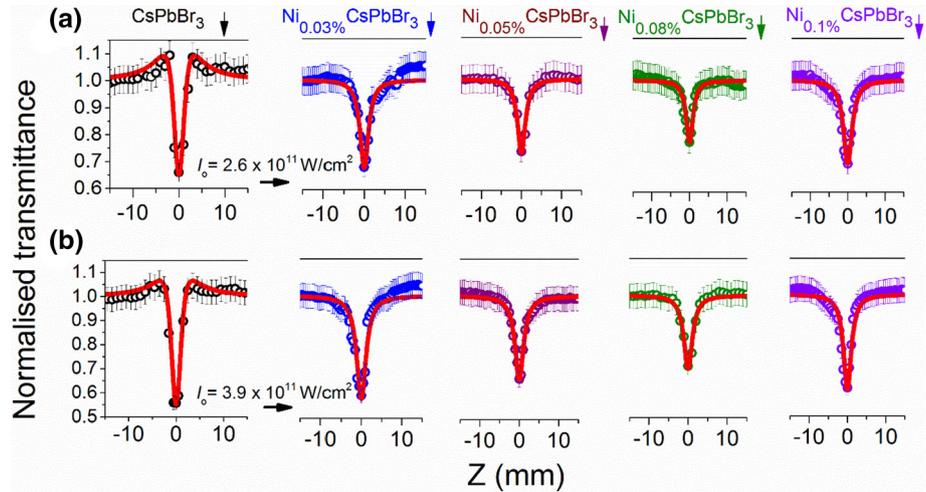
$$\gamma = \frac{\lambda \Delta\Phi_0}{2\pi I_0 L_{\text{eff}}} \quad (4)$$

Figures 3 and 4 show the OA and CA Z-scan curves for  $\text{CsPbBr}_3$  and Ni (0.03, 0.05, 0.08 and 0.1 %)-doped  $\text{CsPbBr}_3$  NCs at input laser intensities  $I_0 = 2.6 \times 10^{11} \text{ W/cm}^2$ , and  $3.9 \times 10^{11} \text{ W/cm}^2$ , respectively. Five measurements were carried out for each OA and CA Z-scans and the average data are presented in the corresponding Figs. 3 and 4.

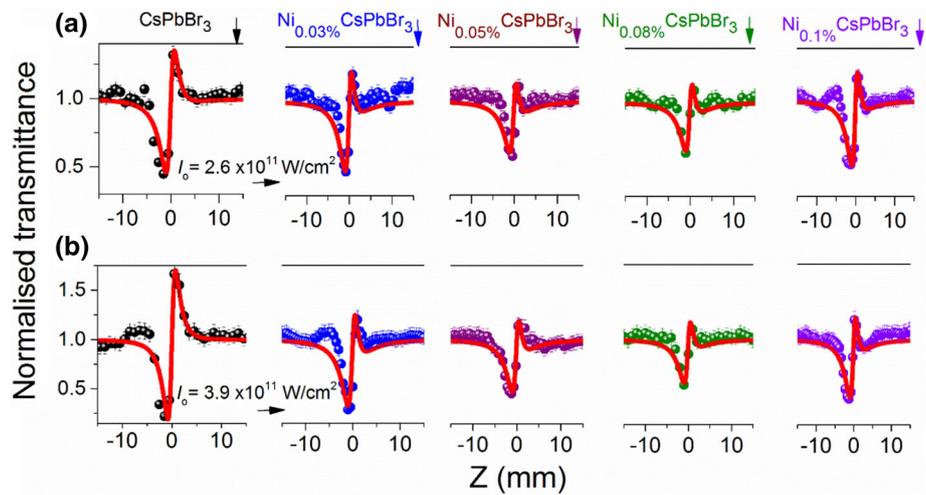
For  $\text{CsPbBr}_3$  NCs, the OA Z-scan curves illustrated an upward trend at the outer part of the focal plane,

suggesting the presence of SA. However, once the sample approached the focus, the Z-scan data depicted a sharp dip in the transmittance representing the appearance of RSA. It was also observed that the RSA increased with the growth of input intensities (Fig. 3,  $\text{CsPbBr}_3$  panels). The corresponding  $I_{\text{sat}}$  for pure  $\text{CsPbBr}_3$  NCs at  $I_0 = 2.6 \times 10^{11} \text{ W/cm}^2$  was estimated to be  $2 \times 10^{11} \text{ W/cm}^2$ . Whereas in the case of Ni-doped  $\text{CsPbBr}_3$  NCs, the Z-scan curves demonstrated RSA and the dip moved in the upward direction compared to pristine  $\text{CsPbBr}_3$  as shown in Fig. 3a, b. We observed that Ni-dopant presence demonstrated a massive impact on the nonlinear absorption process. At the outer part of the focal plane, Ni-counterparts possessed RSA and at the focus they depicted SA, which was confirmed by dividing the pure  $\text{CsPbBr}_3$  contribution from the Z-scans of Ni-doped  $\text{CsPbBr}_3$  NCs. The corresponding divided Z-scan data are shown in Fig. 5a, b. The plots also show that the increment in the percentile

**Figure 3** Open-aperture Z-scan curves in the case of pristine and Ni-doped CsPbBr<sub>3</sub> nanocrystals, at the **a**  $2.6 \times 10^{11}$  and **b**  $3.9 \times 10^{11}$  W/cm<sup>2</sup> intensities of laser pulses. Open symbols represent experimental data points while the solid lines are theoretical fits.



**Figure 4** Closed-aperture Z-scan curves in the case of pristine and Ni-doped CsPbBr<sub>3</sub> nanocrystals at the **a**  $2.6 \times 10^{11}$  and **b**  $3.9 \times 10^{11}$  W/cm<sup>2</sup> intensities of laser pulses. Solid symbols represent experimental data points while the solid lines are theoretical fits.



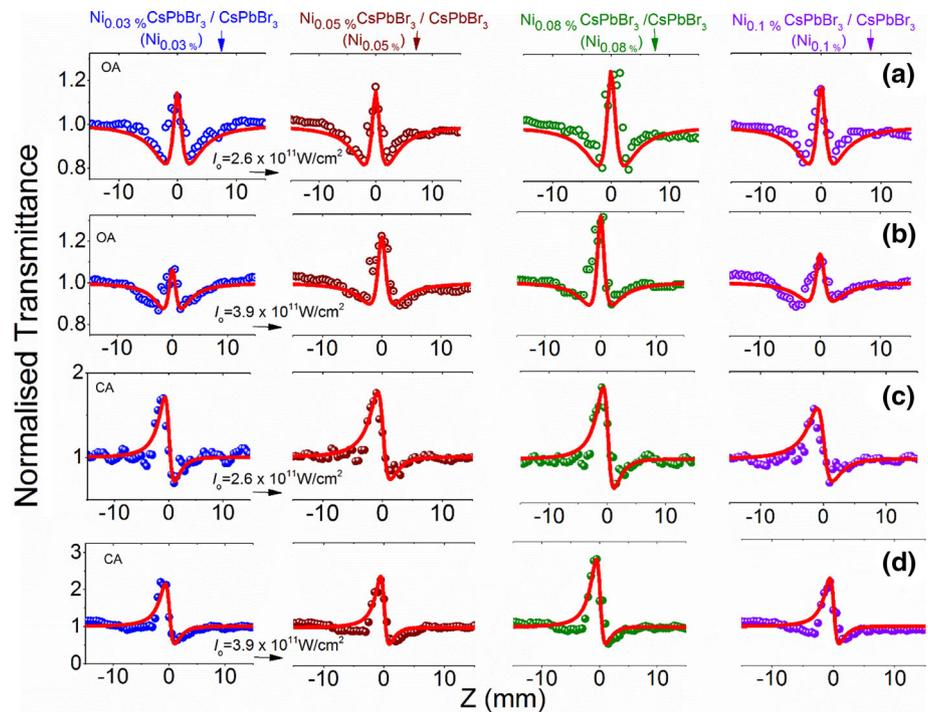
change of Ni-dopants depicts a nonlinear growth of the saturable absorption peaks at focus. The conclusion about the growing role of RSA was also confirmed during the division of the pure CsPbBr<sub>3</sub> contribution from Ni-doped CsPbBr<sub>3</sub> NCs in the case of CA measurements (see data presented in Fig. 5(c, d)).

Pure and doped NCs illustrated a self-focusing behavior, evident from the CA Z-scan data presented in Fig. 4. However, the addition of Ni led to a decrease in self-focusing. Consequently, Ni dopants showed the self-defocusing properties and the nonlinear refractive index of dopants has a negative sign (see Figs. 5(c, d, 6b)]. The experimental data of OA and CA measurements at  $I_0 = 2.6 \times 10^{11}$  W/cm<sup>2</sup> and  $3.9 \times 10^{11}$  W/cm<sup>2</sup> of the undoped and doped NCs are fitted using Eqs. 1, 2 and 3, respectively and the calculated NLO parameters are listed in Table 1. Whereas, for

ratio of NiCsPbBr<sub>3</sub> NCs to the pure CsPbBr<sub>3</sub> NCs the obtained  $I_{sat}$  and  $\gamma$  values are shown in Table 2.

In this work, we have pumped with 400 nm (3.0 eV) pulses for extracting the NLO coefficients of pure CsPbBr<sub>3</sub> and NiCsPbBr<sub>3</sub> NCs. The obtained results demonstrate that the NiCsPbBr<sub>3</sub> NCs possessed RSA and self-focusing properties. However, these processes magnitude were found to be decreased with the growth of the Ni concentration. Consequently, it was observed that for NiCsPbBr<sub>3</sub> NCs possess lower  $\beta$  and  $\gamma$  values compared to pure CsPbBr<sub>3</sub> NCs. This is due to the fact that the Ni dopants possess negative nonlinear absorption coefficient (due to the influence of SA) and negative nonlinear refractive index compared to the pure CsPbBr<sub>3</sub> NCs. Correspondingly, the studied mixture (Ni + CsPbBr<sub>3</sub> NCs) shows smaller values of  $\beta$  and  $\gamma$  compared to the pure CsPbBr<sub>3</sub> NCs. The mechanisms of Ni dopants are extracted from

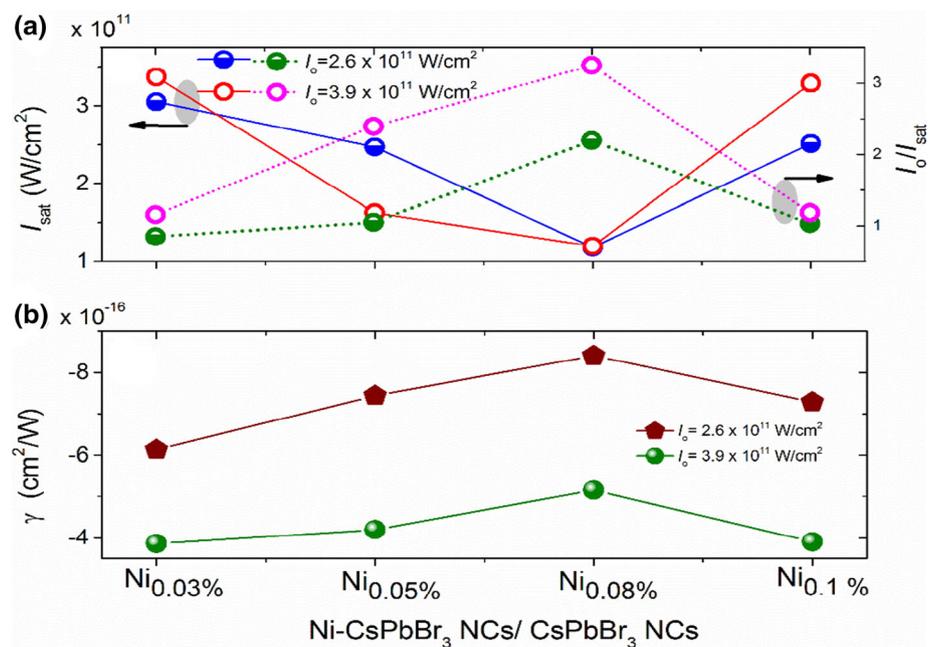
**Figure 5** The ratio of Z-scans at  $2.6 \times 10^{11}$  and  $3.9 \times 10^{11}$  W/cm<sup>2</sup> for the Ni-doped CsPbBr<sub>3</sub> NCs with regard to the pure CsPbBr<sub>3</sub> in the case of **a, b** OA and **c, d** CA measurements. Open and solid symbols represent experimental data points while the solid lines are theoretical fits..



both OA and CA measurements by dividing the pure CsPbBr<sub>3</sub> NCs data to NiCsPbBr<sub>3</sub> data. Meanwhile, after the deduction of pure CsPbBr<sub>3</sub> contribution from NiCsPbBr<sub>3</sub> NCs, though Ni doped samples demonstrated decreased  $I_{\text{sat}}$  values with respect to their % changes. In this case, the materials transmitted higher power than the pure NCs. These materials can be used for optical limiting and/or optical

switching applications. Furthermore, we believe these materials have the potential for optical communications/sensing applications since the nonlinearities can be tuned by changing the dopant and the doping levels. See, for example, a recent report by Gopala Krishna et al. [76] wherein they demonstrated that doping resulted in the passivation of surface defect states, which improved the photoluminescence

**Figure 6** The variations of saturation intensities (left: Y-axis) and ratios of  $I_o/I_{\text{sat}}$  (right: Y-axis) **(a)** and nonlinear refraction indices **(b)** at different concentrations of Ni using two intensities ( $I_o = 2.6 \times 10^{11}$  W/cm<sup>2</sup> and  $3.9 \times 10^{11}$  W/cm<sup>2</sup>).



**Table 1** Summary of the NLO parameters of studied samples at  $\lambda = 400$  nm

Samples	At $I_0=2.6 \times 10^{11}$ W/cm <sup>2</sup>		At $I_0=3.9 \times 10^{11}$ W/cm <sup>2</sup>	
	$\beta(\times 10^{-11})$ cm/W	$\gamma(\times 10^{-16})$ cm <sup>2</sup> /W	$\beta(\times 10^{-11})$ cm/W	$\gamma(\times 10^{-16})$ cm <sup>2</sup> /W
CsPbBr <sub>3</sub>	4.88±0.23	6.6±0.33	6.52±0.32	7.58±0.37
Ni <sub>0.03%</sub> -CsPbBr <sub>3</sub>	4.37±0.21	5.09±0.25	3.83±0.19	4.20±0.21
Ni <sub>0.05%</sub> -CsPbBr <sub>3</sub>	3.59±0.17	3.40±0.17	3.06±0.15	3.27±0.16
Ni <sub>0.08%</sub> -CsPbBr <sub>3</sub>	3.03±0.15	3.30±0.16	2.70±0.13	2.79±0.13
Ni <sub>0.10%</sub> -CsPbBr <sub>3</sub>	4.06±0.20	4.78±0.23	3.56±0.17	3.60±0.18

**Table 2** Summary of the NLO parameters for only Ni NCs. The NLO parameters were obtained from division data of Ni CsPbBr<sub>3</sub> to pure CsPbBr<sub>3</sub> NCs

Samples (%NiCsPbBr <sub>3</sub> / CsPbBr <sub>3</sub> )	At $I_0=2.6 \times 10^{11}$ W/cm <sup>2</sup>		At $I_0=3.9 \times 10^{11}$ W/cm <sup>2</sup>	
	$I_{\text{sat}}(\times 10^{11})$ W/cm <sup>2</sup>	$-\gamma(\times 10^{-16})$ cm <sup>2</sup> /W	$I_{\text{sat}}(\times 10^{11})$ W/cm <sup>2</sup>	$-\gamma(\times 10^{-16})$ cm <sup>2</sup> /W
Ni <sub>0.03%</sub>	3.05	6.13	3.37	3.86
Ni <sub>0.05%</sub>	2.46	7.43	1.62	4.19
Ni <sub>0.08%</sub>	1.18	8.40	1.19	5.16
Ni <sub>0.10%</sub>	2.51	7.29	3.29	3.90

quantum yield and PL lifetime of the CsPbBr<sub>3</sub> nanocrystals. Figure 6 shows the variations of saturation intensities and nonlinear refractive indices of Ni components at the input intensities  $I_0= 2.6 \times 10^{11}$  and  $3.9 \times 10^{11}$  W/cm<sup>2</sup>, respectively. The 0.05 and 0.08% Ni-containing NCs possessed lower saturation intensities than the 0.03 and 0.1% Ni.

We also deduced that with further growth of laser intensity, the NLO coefficients decreased. The nonlinear coefficients ideally should not depend on excitation fluence. It is true for the small intensities of the probe pulses. It is well known in the literature that increasing input intensity will result in further excitation from the higher energy levels (conduction band in this case). Depending on the input peak intensities, a SA process can switch to RSA process and vice-versa [see, for example, Ref. [82] in the case of gold nanoparticles prepared by laser ablation techniques and Ref. [83] in the case of Au coated triangular Ag-Au nanostructures]. When there are excitations into the higher energy states, the NLO coefficients get modified with the presence of other/higher-order NLO processes.

We observed the tendency of NLO coefficients to look similar at two used intensities of laser pulses, i.e., at  $2.6 \times 10^{11}$  W/cm<sup>2</sup> and  $3.9 \times 10^{11}$  W/cm<sup>2</sup>. It is assumed that, at higher input laser intensities, the valence band electrons are trapped in the higher

levels of the conduction band. Probably this led to a slight decrease in the NLO coefficients. Moreover, multiple factors contribute to the NLO response of the materials. In brief, the influence of the pumping wavelength, input pulse duration, and the sizes/shapes of nanoparticles/quantum dots/nanocrystals significantly affect the NLO parameters. Zhu et al. reported  $\beta$  and  $\gamma$  values of CsPbBr<sub>3</sub> QDs using excitation wavelengths of 700–1200 nm at 30 kHz repetition, 34–172 fs pulses, and they reported that  $\beta$  and  $\gamma$  were considerably different with respect to different excitation wavelengths [84]. Analysis of size-dependent off-resonant nonlinear optical properties of gold nanoparticles showed the variation of the sign of nonlinear refraction index and nonlinear absorption coefficient for different Au NPs [85]. The same can be said about the influence of pulse duration of the probe radiation on the nonlinear optical response of nanoparticles [86]. We have systematically compared the NLO parameters at different excitation parameters for different sizes and shapes as of CsPbBr<sub>3</sub> perovskites reported by other research groups, as shown in Table 3.

The typical energy level diagrams of CsPbBr<sub>3</sub>, as well as Ni-doped CsPbBr<sub>3</sub> NCs, are illustrated in Fig. 7. We believe that the transitions responsible for RSA are different for Ni-doped CsPbBr<sub>3</sub> NCs compared to pristine CsPbBr<sub>3</sub> NCs because of the

modification of their energy level structure due to doping. It is known that Ni doping reduces the bandgap and induces the defect states in the bandgap region (closer to the conduction band). Possibly, with a higher doping percentage, the density of defect states also increases. Furthermore, at  $\lambda = 400$  nm, the absorbance in doped NCs is higher than that in pure NCs. Once the electrons are excited to the conduction band (with 400 nm pump photon corresponding to  $\sim 3.0$  eV energy), the saturation effect dominates at lower peak intensities since the NCs possess strong linear absorption. However, with increasing peak intensities, there is a possibility of RSA (as shown in Fig. 7).

In pure NCs, the RSA is dominant even at lower peak intensities. The pertinent issue here is the depopulation from excited states to either the lower part of the conduction band (pure case) or to either lower part of the conduction band followed by a transition to the defect states (in doped case), which will possibly affect further absorption leading to either stronger RSA or slightly weaker RSA, depending on the possible scenarios, and thereby influencing the  $I_{\text{sat}}$ . The lifetimes (intra-band relaxation) of the electron, typically in the femtosecond time domain, in the excited states are significant for further absorption through RSA. Further detailed high-resolution pump-probe investigations are necessary to explore this effect.

We estimated the bandgap of  $\text{CsPbBr}_3$  as 2.34 eV (Fig. 7a), which is close to the one reported in Ref. [87]. The doped divalent element can form the defect levels in the perovskite structure, which trap the electrons and reduce the recombination of charge carriers [71].  $\text{Ni}^{2+}$  has partially filled d-orbital and thus can create the donor levels in the bandgap of  $\text{CsPbBr}_3$ , leading to a decrease in bandgap (Fig. 7b). The nonlinear absorption mechanism depends on the material's bandgap and pumping photon energy. The initial absorption process for an electron moving from valence to conduction band can be quantified by considering the ratio between material bandgap and pump photon energy. It is estimated that the ratio values equal to one or less shows saturable absorption, i.e., the case when the pump photon energy is equal or higher than the bandgap of material leads to SA. However, the total nonlinear absorption process depends on the intensity of photon energy and energy levels of the conduction band. In the current case, pure  $\text{CsPbBr}_3$  NCs demonstrated

SA at lower intensities (out of focal plane). When intensity of pulses reached peak values, the RSA is predominated (for Z-scan results, please see Fig. 3). Similarly, if the photon energy is lower than the bandgap of a material then it leads to a multi-photon absorption process. Depending on pump photon energy, the initial absorption process may lead to 2PA, 3PA, 4PA, and 5PA, etc. For example, the ratio values for 2.34 eV to pump photon energy between 700–1200 nm range; the absorption process agrees with previous work reported by Chenyang Zhu et al. [84], i.e., at 700–1100 nm the  $\text{CsPbBr}_3$  QDs possess 2PA and at 1100–1200 nm having 3PA.

In the case of  $\text{NiCsPbBr}_3$ , the addition of Ni probably led to a decrease in the bandgap (Fig. 7b). Therefore, at the peak intensities of pump wavelength, Ni NCs possessed SA and the saturation varies depending on the concentration of Ni NCs. It was observed that the absorption process for pure NCs is SA+RSA. In the case of Ni NCs, near to the focus, they possess SA; due to this effect near the focus, there is a decrease in the RSA compared to pure NCs (see Fig. 3). In conclusion, the NLO process could be influenced by pumping wavelength and pulse duration and the size and shape of the small-sized species. Therefore, it is necessary to explore the NLO properties of materials at required pump wavelengths along with different pulse durations.

In the case of magnetic properties of the synthesized materials, the electron and hole are subjected to various Zeeman interactions in the presence of a magnetic field. As a result, the combined electron-hole energy level broadens and changes, affecting the PL line width appropriately [88]. The presence of a magnetic field results in a net increase in triplet excitons due to the formation of anti-parallel (singlet state) and parallel (triplet state) spin states excitons. When charge-separated states are achieved, then more electrons and holes with parallel spin states will be formed [89]. An elaborative study about the magnetic properties of the synthesized material has to be explored further.

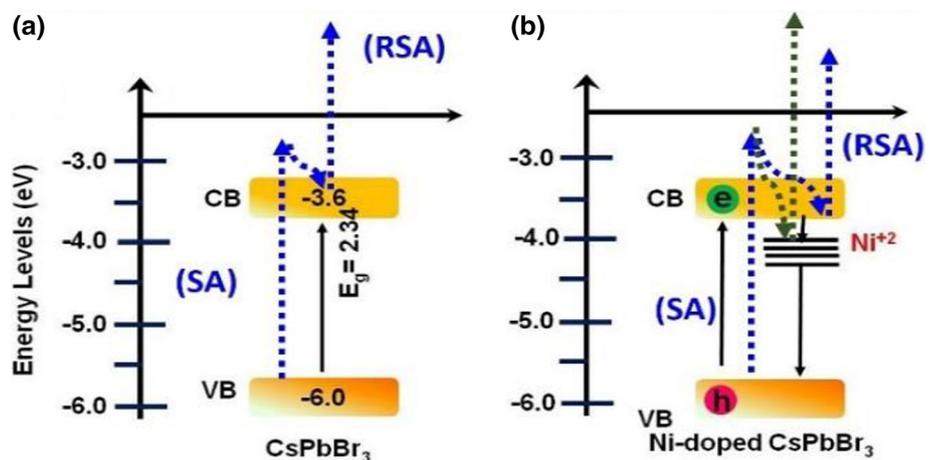
### HHG from LIPs containing nanocrystals using chirp-free 35 fs and chirped 135 fs pulses

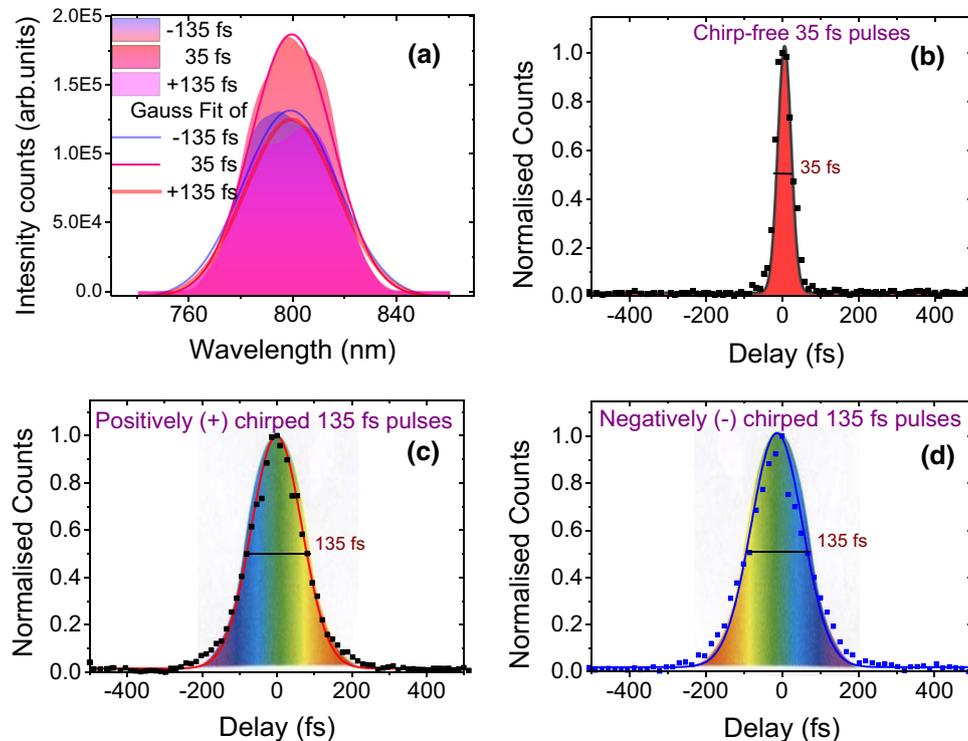
The low-order NLO properties of Ni-doped  $\text{CsPbBr}_3$  NCs at 400 nm wavelength can be used in different applications such as light-emitting diodes,

**Table 3** Summary of the ultrafast NLO properties/coefficients of various perovskite materials

Samples	Size distribution	Excitation details	NLO process	$\beta$ (cm/GW)	$I_{\text{sat}}$ (GW/cm <sup>2</sup> )	$\gamma$ (cm <sup>2</sup> /W)	Ref.
<i>CsPbBr<sub>3</sub> materials</i>							
Square NCs	~ 20 nm	800 nm, 70fs, 1 kHz	2PA	$3.9 \times 10^{-2}$		$3.52 \times 10^{-12}$	[17]
Cubic NCs	~ 9 nm	800 nm, 70 fs, 1 kHz	2PA	0.097			[25]
Nano cubes:	~ 12.4 nm	800 nm, 50 fs, 1 kHz	2PA	0.091			[21]
QDs: spherical (blue emissive)	5 nm	800 nm, 130 fs, 76 MHz	2PA	- 1.71		- $5.18 \times 10^{-15}$	
QDs: nano cubes (green emissive)	~ 17 nm	800 nm, 130 fs, 76 MHz	2PA	- 0.68	0.011	$1.99 \times 10^{-15}$	[32]
Nanosheet:	104.6 nm	800 nm; 80 fs	2PA	10.94	3.12		[77]
	195.4 nm	800 nm; 80 fs	2PA	4.73	4.79		[29]
Mono-crystal	1000 nm	30 ps, 50 Hz	2PA	5.00			[78]
Nanocrystal film:		1060 nm, 500 ps	SA		10.7		[79]
Nano cubes:	10-16 nm	1064 nm, 1 ns, 20 kHz	3PA	$0.9-1.1 \text{ cm}^3/\text{GW}^2$	0.8-1	$0.8-1.2 \times 10^{-12}$	[18]
Single crystal:		1200 nm	3PA	$\gamma = 0.14 \text{ cm}^3/\text{GW}^2$			[79]
		2050-2300 nm, 50 fs.	5PA	$\eta\delta_5 = (0.09-6.5) \times 10^{-136} \text{ cm}^{10} \text{ s}^4 \text{ photon}^{-4}$			[18]
<i>Other perovskites</i>							
QDs cubes:	~ 12.4 nm	800 nm, 50fs, 1 kHz	2PA	0.054			
CsPbCl <sub>2</sub> Br:				0.064			[21]
Cs <sub>2</sub> AgIn <sub>0.9</sub> Bi <sub>0.1</sub> Cl NCs	1-5nm	800 nm, 57fs	2PA	6.88		$55 \times 10^{-13}$	[80]
CsPbI <sub>1.5</sub> Br <sub>1.5</sub>	~ 40 μm	1064 nm, 2 ps, 58 MHz	2PA	0.94			[81]

**Figure 7** Energy levels of **a** CsPbBr<sub>3</sub> and **b** Ni-doped CsPbBr<sub>3</sub> nanocrystals. Also depicted are the transitions for RSA and SA (blue dotted lines).





**Figure 8** Spectral shapes and spectro-temporal distributions of used laser pulses for HHG. **a** Spectra of chirp-free 35 fs pulses, negatively chirped 135 fs pulses, and positively chirped 135 fs pulses. **b** Autocorrelation trace of 35 fs pulses. Spectral distribution along these chirp-free pulses represents the homogeneously distributed red and blue components in both the leading and trailing parts of the laser pulse. **c** Autocorrelation trace

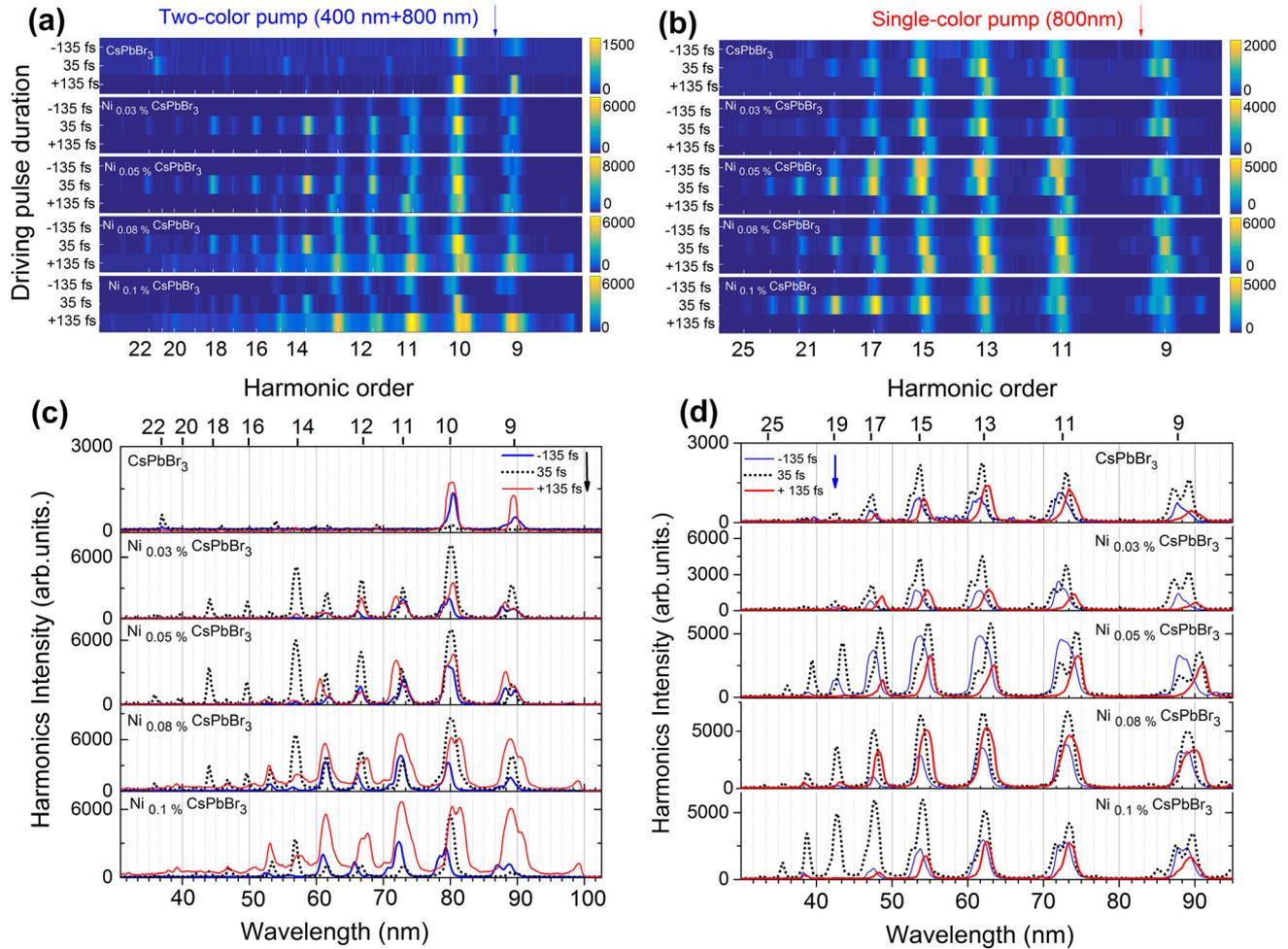
and photodetectors, and lasers. Meanwhile, in our earlier work, we have already demonstrated the application of such species for HHG of the chirp-free 35 fs pulses using the same samples, which were ablated by nanosecond and picosecond heating pulses [73]. In this subsection, we analyze the harmonics emission from LIPs of reported NCs using chirped TCP and SCP driving laser pulses and compare the results with chirp-free pulses.

The spectral profiles of chirp-free and chirped pulses were measured using a USB spectrometer (Ocean Optics) and are shown in Fig. 8a. All these pulses possess the same spectral components. Meanwhile, their distribution along the laser pulses significantly differs from each other. The blue and red components were equally distributed along the temporal shape of chirp-free 35 fs pulses (Fig. 8b). The spectral distribution along the negatively and positively chirped 135 fs pulses is shown in Fig. 8c, d, respectively. The blue and red components are

and spectral distribution along the positively chirped 135 fs pulses. The red component is moved toward the leading front of the pulse. **d** Autocorrelation trace and spectral distribution along the negatively chirped 135-fs pulses. The blue component is moved toward the leading front of the pulse. The squares shown in (b–d) represent the autocorrelation traces. Solid lines represent the Gaussian fits of experimentally obtained spectra.

moved toward the leading front and the tailing part of the pulses, respectively, for negatively chirped pulses. Whereas, for positively chirped 135 fs pulses, the reverse pattern is seen compared with the negatively chirped pulses. The red component is moved toward leading front of pulse, while blue component of spectrum is concentrated at the tailing part of the laser pulse.

Figure 9a, b show 2D color map and Figure 9c, d depict the corresponding line profiles of HHG spectra from 2D NCs generated by two-color pump and single-color pump of LIPs using chirp-free 35 fs pulses, negatively and positively chirped 135 fs pulses, respectively. The intensities of chirped pulses during propagation through the LIPs were  $0.9 \times 10^{14}$  W/cm<sup>2</sup>. The laser intensity of the chirp-free 35-fs pulse was  $4.0 \times 10^{14}$  W/cm<sup>2</sup>. Similar pulse energies (0.6 mJ) of the driving pulses were used in the case of chirp-free and chirped pulses. The harmonics shifted toward shorter and longer wavelengths in the case of



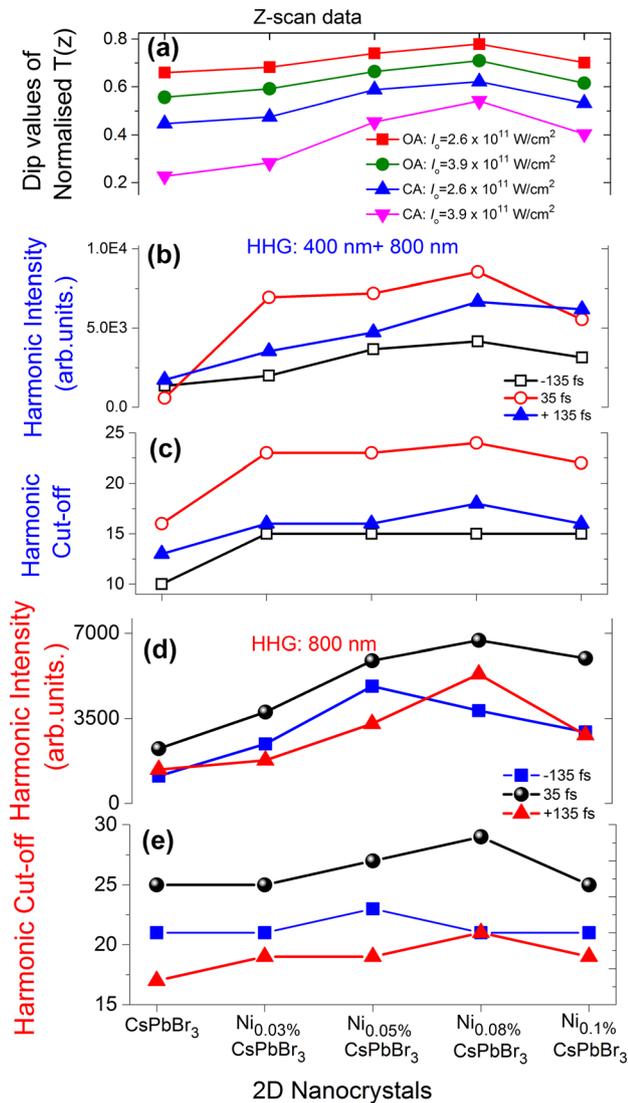
**Figure 9** a, b 2D color map and c, d corresponding line profiles of Harmonic spectra from 2D nanocrystals generated by two-color pump and single-color pump of LIPs using chirp-free 35 fs pulses, negatively and positively chirped 135 fs pulses, respectively.

negatively and positively chirped driving laser pulses, respectively. For example, tenth harmonic (10H) in the case of TCP and ninth harmonic (9H) in the case of SCP of plasma showed  $\sim 0.30$  and  $\sim 0.85$  nm blue shifts, in the case of negatively chirped driving laser pulses with regard to the harmonics generated using chirp-free pulses. The leading parts of negatively and positively chirped DPs produced the harmonics from the blue and red spectral components, respectively, which led to the blueshift and redshift of harmonics. Therefore, one can finely control the shift of harmonics by changing the chirp of driving laser pulses. These modifications of harmonic spectra can find application in nonlinear spectroscopy of materials using tunable coherent XUV pulses.

The cut-off of harmonics is reduced for chirped pulses due to a decrease in the peak intensity of DPs. In the case of SCP, the chirp-free pulses produced

maximum harmonic cut-off compared with chirped pulses. Also, for all samples, the even harmonics have higher intensities than odd harmonics. Higher intensity of the harmonics is attributed to the shorter wavelength sources (400 nm) compared with the 800 nm radiation, which is related to the growth of ponderomotive energy ( $U_p$ ) of DP. In other words, the enhancement of harmonic yield is proportional to  $\lambda^{-5}$  or  $\lambda^{-6}$  [90–92]. Whereas, the cut-off of harmonics is proportional to  $\sim \lambda^2$ , since the energy cut-off is given by  $E_{\text{cut-off}} = I_p + 3.17U_p$ . Where  $I_p$  is the ionization potential,  $U_p = 9.33 \times 10^{-14} I$  (W/cm<sup>2</sup>)  $\lambda^2$  ( $\mu\text{m}$ ), and  $I$  is the laser intensity [93].

Below we summarize our studies. The addition of Ni-dopants to the CsPbBr<sub>3</sub> NCs demonstrated significant changes in the NLO properties at different excitation wavelengths. In our earlier work [17] (800 nm, Z-scan measurements), the nonlinear absorption



**Figure 10** Summary graphs of the NLO response of studied samples versus the concentration of Ni. **a** Valley values of normalized  $T(z)$  for OA and CA measurements at 400 nm. **b–e** Harmonics intensity and cut-off for TCP and SCP chirp free and chirped driving pulses (see text).

coefficients and nonlinear refractive indices are increased with % change of Ni compared to the pure NCs. In contrast, in the current case with excitation by 400 nm pulses, the NLO coefficients are decreased. However, in this case the Ni dopants possess a self-defocusing effect, and in both cases, i.e., for 400 nm (only Ni contribution, current work) and 800 nm (earlier work [17]), we observed the incremental variation of  $\gamma$ . In both cases, the incremental trend is almost similar to HHG yields from the same studied samples for TCP (800 nm+400 nm) and SCP (800 nm).

In the case of HHG using TCP, the harmonic generation originates from both excitation wavelengths, i.e., from 800 nm and 400 nm pumps. Therefore, we have shown that higher nonlinear refractive indices for Ni-doped NCs correlate with higher HHG yields from the plasma plumes composed of the components of Ni+CsPbBr<sub>3</sub>. Adding Ni to pure CsPbBr<sub>3</sub> could enhance the NLO response in terms of  $\gamma$  for 800 nm and 400 nm measured using Z-scan technique, and a similar behavior is also achieved for HHG yields for both driving schemes (SCP and TCP).

Meanwhile, by considering the nonlinear absorption, the Ni dopants demonstrate SA at higher intensities. In Fig. 6a, we showed the calculated  $I_{\text{sat}}$  and  $I_0/I_{\text{sat}}$  ratios for Ni NCs. The  $I_0/I_{\text{sat}}$  and  $\gamma$  dependencies look similar. Figure 10a shows the values of valleys in the OA and CA normalized transmittances; these curves are similar to the HHG curves for the samples shown in Figs. 10b–e. Finally, we can summarize that the NLO coefficients measured by Z-scan technique using 800 and 400 nm excitation wavelengths and HHG yields demonstrate similar tendency with variation of the Ni concentration.

## Conclusions

In this work, we have explored the third-order NLO properties of pure CsPbBr<sub>3</sub> and Ni-doped CsPbBr<sub>3</sub> NCs using 400 nm, 60 fs, 1 kHz laser pulses. We also demonstrated the enhanced high-order harmonics generation in the plasmas produced on the surfaces of the latter species compared with the pristine NCs using different configurations of the driving pulses. The OA Z-scan measurements have shown that pure CsPbBr<sub>3</sub> NCs possess SA and RSA at lower and higher intensities of laser pulses, respectively. These samples have shown a self-focusing effect. However, the nonlinear absorption and refraction decreased for Ni-doped CsPbBr<sub>3</sub> NCs due to the opposite NLO properties of Ni dopants with regard to the CsPbBr<sub>3</sub> due to the growing influence of the SA and self-defocusing attributed to the Ni. In addition, we have examined the high-order harmonics emission from the LIPs of these NCs using TCP and SCP chirp-free and chirped driving laser pulses. The Ni-dopants have increased the harmonic cut-off and yield from the corresponding plasmas. We have also demonstrated the advantages of chirped laser pulses for

tuning the harmonics, which might be helpful for the XUV nonlinear spectroscopy of different materials. In summary, our detailed studies have demonstrated that  $\beta$  and  $\gamma$  values for combined NiCsPbBr<sub>3</sub> NCs are decreased compared to pure CsPbBr<sub>3</sub> NCs. However, if we consider NLO coefficients for Ni NCs, the  $\gamma$  values are increased with the growth of Ni doping concentration. The same tendency was observed in the harmonics yield and cut-off for both SCP and TCP of LIPs. Thus, the correlation between the low-order NLO properties and HHG of similar materials was demonstrated.

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## Authors contributions

SRK contributed to conceptualization, methodology, writing—original draft, writing—review and editing. VRS contributed to conceptualization, writing—review and editing. RAG contributed to conceptualization, writing—review and editing. MB and RK contributed to synthesis of samples, characterization of nanocrystals. WL contributed to conceptualization, supervision, funding acquisition.

## Data availability

The data that support the findings presented here are available from the authors upon reasonable request.

## Declarations

**Conflict of interest** The authors declare no conflicts of interest.

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